

PHAEOPHYTINS FROM *Thyrsacanthus ramosissimus* Moric. WITH INHIBITORY ACTIVITY ON HUMAN DNA TOPOISOMERASE II- α [#]

Analúcia Guedes Silveira Cabral, Fábio Henrique Tenório-Souza, Marcelo Dantas Moura, Sabrina Gondim Ribeiro Mota, Antônio Cláudio da Silva Lins, Celidarque da Silva Dias e José Maria Barbosa-Filho*

Departamento de Ciências Farmacêuticas, Universidade Federal da Paraíba, CP 5009, 58051-970 João Pessoa – PB, Brasil

Ana Maria Giulietti

Departamento de Ciências Biológicas, Universidade Estadual de Feira de Santana, 44036-900 Feira de Santana – BA, Brasil / Royal Botanic Gardens, Kew, TW9 3AB, UK

Tania Maria Sarmiento da Silva

Departamento de Ciências Moleculares, Universidade Federal Rural de Pernambuco, 52171-900 Recife – PE, Brasil

Creusioni Figueredo dos Santos

Departamento de Biologia Molecular, Universidade Federal da Paraíba, 58051-900 João Pessoa – PB, Brasil

Recebido em 22/5/12; aceito em 21/7/12; publicado na web em 15/10/12

Our study reports the extraction and isolation of a new phaeophytin derivative 15¹-hydroxy-(15¹-5)-porphyrinolactone, designated anamariaine (**1**) herein, isolated from the chloroform fraction of aerial parts of *Thyrsacanthus ramosissimus* Moric. along with the known 15¹-ethoxy-(15¹-5)-porphyrinolactone (**2**). These compounds were identified by usual spectroscopic methods. Both compounds were subjected to *in vitro* (inhibitory activity) tests by means of supercoiled DNA relaxation techniques and were shown to display inhibitory activity against human DNA topoisomerase II- α at 50 μ M. Interconversion of these two pigments under the mild conditions of the isolation techniques should be highly unlikely but cannot be entirely ruled out.

Keywords: *Thyrsacanthus ramosissimus*; phaeophytins; topoisomerase activity.

INTRODUCTION

The family Acanthaceae comprises around 250 genera and approximately 2500 species,¹ which occur mainly in the Atlantic Forest and in the mesophilic forest formations of the Central-Western and Southeastern regions of Brazil, as well as in other types of vegetation.² Previous studies on the isolation of natural products from the family Acanthaceae have revealed the presence of alkaloids,³ flavonoids,⁴ terpenes,⁵ coumarins,⁶ lignans,⁷ and quinoids.⁸ These constituents have been demonstrated to have biological effects such as cytotoxicity,⁷ vasorelaxant,⁵ anti-inflammatory,⁹ antifungal,⁵ and antiviral actions,¹⁰ as well as CNS depression and immunosuppressive activities.^{9,11}

Thyrsacanthus Moric. is a genus with five species whose presence is restricted to South America and that occurs mainly in Brazil. Recently, the genus was restored to include the South American species traditionally assigned to *Anisacanthus* Nees. *Thyrsacanthus ramosissimus* Moric. (= *Anisacanthus brasiliensis* Lindau) is popularly known as “canudo” and is endemic to Brazil, in the states of Alagoas, Bahia, Minas Gerais, Pernambuco, and Rio Grande do Norte. It can be found principally in seasonally dry vegetation and in semideciduous and riparian forests.¹² No biological studies have been previously performed with this species and chemical and biological studies with species of this genus are also scarce.

Interest in compounds with inhibitory activity against the enzyme DNA-topoisomerase II- α has increased,¹³⁻²⁰ because intracellular levels of topoisomerase II- α (topo II- α) are elevated in a number of human tumors as compared to the respective normal tissues. Moreover, the development of drugs that can affect the DNA replication process by selectively interfering with the function of topo II- α continues to draw researchers' attention.²¹ Topo II- α is an essential enzyme that

plays a key role in DNA replication, but is also important in repair, transcription, and chromosome segregation. Topo II- α changes DNA topology by passing an intact DNA double helix through a transient double-stranded break, thereby producing another helix.^{22,23} Thus, as part of our ongoing chemical investigations of this plant species,²⁴ in this work we evaluate the inhibitory effect of this plant on human DNA-topoisomerase II- α . The isolation of a new phaeophytin (**1**), as well as its structural elucidation by means of ESI-MS and 1D- and 2D-NMR experiments, will be reported here.

EXPERIMENTAL

General experimental procedures

Melting points were determined on a Koeffler hot stage and are uncorrected. The infrared absorption spectra were recorded in KBr pellets, using a Bomem/MB-102 spectrophotometer operating in the 4000-400 cm^{-1} range. The LC-MS spectra were obtained in the positive electrospray mode using a Quattro LC-Micromass device (Waters). Silica gel 60 was used for column chromatography, and Kieselgel 60F₂₅₄ (E. Merck) was employed for preparative TLC as precoated plates. Sephadex LH-20 (Sigma) was utilized for gel permeation chromatography. ¹H and ¹³C NMR spectra were acquired on a Bruker AC 200 spectrometer (200 MHz for ¹H and 50 MHz for ¹³C), in CDCl₃.

Plant material

The aerial parts of *Thyrsacanthus ramosissimus* Moric. were collected in the city of Rio de Contas, state of Bahia, Brazil, in March 2006. The plant was identified by Dr. A. M. Giulietti. A voucher specimen (Tombo HUEFS 59791) was deposited at the Herbarium of Universidade Estadual de Feira de Santana.

*e-mail: jbarbosa@lft.ufpb.br

[#]Artigo em homenagem ao Prof. Otto R. Gottlieb (31/8/1920-19/6/2011)

Extraction and isolation

The powdered plant material (5.0 kg) was extracted successively with EtOH, to give 310.0 g dry extract. This extract was dissolved in MeOH/H₂O (3:7) and successively fractionated with hexane, CHCl₃, and AcOEt, yielding the hexane (20.0 g), CHCl₃ (10.0 g), and EtOAc (5.0 g) fractions. The CHCl₃ fraction was successively submitted to column chromatography using silica gel and Sephadex LH-20, followed by preparative TLC on silica, to yield **1** (29.2 mg) and **2** (27.8 mg) as dark blue amorphous solids. Analysis of the spectroscopic data (IR, LC-MS and NMR spectra, including 2D NMR) led to identification of the following compounds: anamariaine (15¹-hydroxy-(15¹-*S*)-porphyrinolactone, (**1**) and 15¹-ethoxy-(15¹-*R*)-porphyrinolactone (**2**).

Anamariaine (**1**)

Dark blue amorphous solid; mp 171.0 °C; IR (KBr) ν_{\max} 3549,

3476, 3414, 1737, 1638, 1615 cm⁻¹. ¹H NMR (CDCl₃, 200 MHz) (Table 1), ¹³C NMR (CDCl₃, 50 MHz) (Table 1), ESI-MS (pos) *m/z* 903.99 [M+H]⁺ (C₃₅H₇₄N₄O₇).

15¹-Ethoxy-(15¹-*R*)-porphyrinolactone (**2**)

Dark blue amorphous solid; mp 114.0 °C; IR (KBr) ν_{\max} 3447, 2927, 1734, 1636, 1104 cm⁻¹, ESI-MS (pos) *m/z* 931.59 [M+H]⁺ (C₃₇H₇₉N₄O₇).

In vitro assay for DNA topoisomerase II- α

The conversion of pBR322 supercoiled plasmid DNA to the relaxed form by topo II- α was examined in the presence of phaeophytins **1** and **2**. DNA topoisomerases are enzymes that modulate the topological state of DNA and are targets for many active drugs in cancer treatment.^{15,23}

Enzymatic activity was analyzed by the DNA relaxation assay

Table 1. ¹H (200 MHz) and ¹³C (50 MHz) spectral data for anamariaine (**1**) obtained by heteronuclear 2D shift-correlated HSQC, HMQC, and HMBC spectra, in CDCl₃. Chemical shifts (δ , ppm) and coupling constants (*J* in Hz, in parenthesis)^a

Porphyrin moiety			Phytol moiety		
Position	δ_c	δ_H	Position	δ_c	δ_H
1	141.18		P1	61.45	4.52 (m)
2	131.45		P2	117.70	5.20 (t, 7.0)
2 ¹	12.13	3.42 (s)	P3	142.86	
3	135.98		P3 ¹	16.27	1.60 (s)
3 ¹	128.93	8.05 (dd, 18.0; 12.0)	P4	39.77	1.93 (m)
3 ²	122.71	(<i>E</i>) 6.37 (dl, 18.0) (<i>Z</i>) 6.24 (dl, 7.8)	P5	24.95	1.30 (m)
4	135.98		P6	29.69	1.30 (m)
5	99.59	9.68 (s)	P7	32.73	1.55 (m)
6	155.71		P7 ¹	19.71	0.84 (d, 6.6)
7	136.46		P8	37.36	1.04 (m)
7 ¹	11.24	3.22 (s)	P9	24.39	1.30 (m)
8	145.51		P10	36.59	1.21 (m)
8 ¹	19.52	3.82 (m)	P11	32.58	1.35 (m)
8 ²	17.58	1.68 (t, 8.0 Hz)	P11 ¹	19.71	0.81(d, 6.6)
9	149.94		P12	37.28	1.24 (m)
10	104.10	9.91 (s)	P13	24.76	1.30 (m)
11	138.70		P14	39.32	1.48 (m)
12	134.75		P15	27.94	1.52 (m)
12 ¹	12.43	3.87 (s)	P15 ¹	22.70	0.88 (d, 6.6)
13	111.29		P16	22.70	0.88 (d, 6.6)
13 ¹	166.30				
14	135.98				
15	100.43				
15 ¹	101.95				
15 ²	170.87				
15 ³	54.15	3.75 (s)			
16	161.06				
17	53.68	4.16 (d, 8.7)			
17 ¹	32.13	17 ¹ B 1.81 (m) 17 ² A 2.63 (m)			
17 ²	31.29	17 ² B 2.22 (m) 17 ² A 2.49 (m)			
17 ³	173.30				
18	50.12	4.51 (m)			
18 ¹	22.60	1.62 (d, 8.0 Hz)			
19	171.12				
20	93.87	8.86 (s)			

^a 2D homonuclear ¹H-¹H-COSY and heteronuclear HMBC spectra were also used for these assignments. Hydrogen atoms chemical shifts obtained from ¹D ¹H NMR spectra. Carbon signals corresponding to C, CH, CH₂, and CH₃ as deduced by comparative analysis of the APT and HSQC spectra. Superimposed ¹H signals are described without multiplicity and chemical shifts were deduced by HSQC, HMBC, and ¹H-¹H-COSY.

according to the protocol described by USB Corporation (USB Corporation, Cleveland, OH, USA). One unit of topo II- α (USB Corporation, Cleveland, OH, USA) was incubated with 0.152 μg pBR322 DNA (human recombinant from *E. coli*, Invitrogen) and with 100 or 50 μM of compound **1** or **2**, or without the test compounds, in 10 μL reaction mixture containing 10 mM Tris, pH 7.9, 50 mM NaCl, 50 mM KCl, 5 mM MgCl_2 , 0.1 mM EDTA, 15 $\mu\text{g}/\text{mL}$ BSA, 1 mM ATP, 10 mM Na_2HPO_4 , and 0.2 mM DTT for 40 min, at 37 $^\circ\text{C}$. The reaction was terminated by addition of 1 μL stop solution consisting of 50% glycerol, 10% sodium dodecyl sulfate (SDS), and 25% bromophenol blue. Electrophoresis was carried out on 1% agarose gel (Sigma-Aldrich) equilibrated with TAE buffer (4.84 g L^{-1} Tris-base, pH 8.5, 1.14 g L^{-1} glacial acetic acid and 100 mL of 0.74 g L^{-1} EDTA) for 120 min at 40 V. Etoposide was used as the positive control. The gels were stained with ethidium bromide solution (5 g L^{-1}) after electrophoresis for 30 min, washed with water, and photographed under UV light with a digital camera.

RESULTS AND DISCUSSION

Extensive chromatography of the CHCl_3 fraction of *Thyracanthus ramosissimus* resulted in the isolation and characterization of two phaeophytins **1** and **2** (Figure 1). Compound **1** was isolated as a dark blue amorphous solid. ESI-MS analysis in the positive ionization mode resulted in $[\text{M}+\text{H}]^+ m/z$ 903.99, indicating 21 degrees of insaturation for a molecular formula of $\text{C}_{55}\text{H}_{74}\text{N}_4\text{O}_7$. The IR absorptions at 3549, 3476, 3414, 1737, and 1615 cm^{-1} and UV data, and overall color characteristics suggested the presence of a large chromophore related to the chlorophyll derivative containing a hydroxyl group.

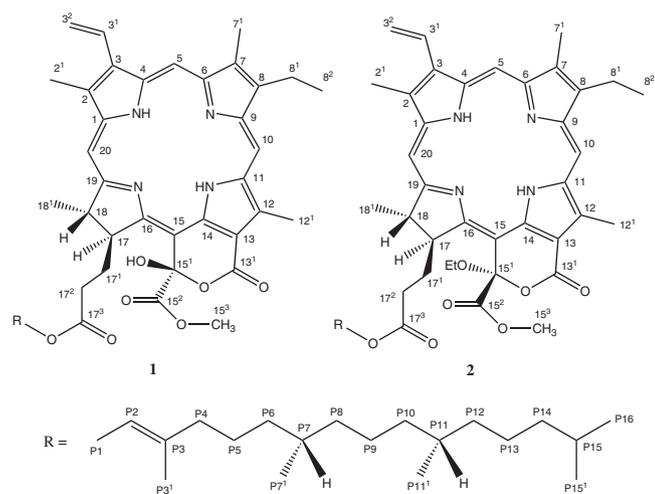


Figure 1. Structures of anamariaine (**1**) and 15¹-ethoxy-(15¹-R)-porphyrinolactone (**2**)

The 200 MHz ¹H NMR spectrum displayed all the resonances of phaeophytin A and its derivatives:²⁵ three aromatic methyl groups (δ_{H} 3.95, 3.48, and 3.33), a methoxyl group (δ_{H} 3.79), three singlet olefinic protons (δ_{H} 9.91, 9.68, and 8.86), vinyl substitution (δ_{H} 6.37, 6.24, and 8.05) with a characteristic exomethylene coupling pattern, and two singlets at δ_{H} 1.50 and 1.41 indicative of two pyrrole NH functionalities within the shielding ring current.²⁶ The attached phytol moiety was recognized by a large number of overlapping proton signals of aliphatic methylene and methyl functions. Proton resonances suggesting the phytol moiety were the carbinol resonances CH_2 -P1 (δ_{H} 4.51), the olefinic proton H-P2 (δ_{H} 5.20) and CH_3 -P3¹ (δ_{H} 1.6) (Figure 1).

Thirty-five carbon signals were detected in the ¹³C NMR spectrum for the porphyrin system of compound **1**, while the aliphatic phytol

moiety afforded a partial overlapping set of signals in the aliphatic region. Overall, the ¹H and ¹³C-NMR spectra were in accordance with the literature data.^{27,28}

The presence of the hydroxyl group in ring E was deduced from the carbinol resonance C-15¹ at δ_{C} 101.95, with a quaternary nature confirmed by the APT and HSQS experiment. The NMR and ESI-MS data, together with the degrees of insaturation, allowed us to establish a conjugated d-lactone structure, which appeared to be formed between the C-13¹-C-13² bond at ring E of 13²-hydroxyphaeophytin A. There was evidence of a carbonyl carbon at δ_{C} 166.30 instead of the resonances at δ_{C} 192.2 for the corresponding carbon.²⁹

The complete analysis of the porphyrin structure was conducted on the basis of the correlations detected in the ^{2,3}J_{CH} long-range of the olefinic protons H-5, H-10, H-20, which provide the connectivities between the four pyrrole rings (Figure 2). The signal corresponding to H-5 (δ_{H} 9.68) correlated with C-7 in ring B. Similarly, H-10 (δ_{H} 9.91) presented connectivity with C-8 (ring B) and C-11 in ring C, while H-20 (δ_{H} 8.86) correlated with C-1 and C-2 (ring A). The signal relative to 3H-18¹ (δ_{H} 1.62) in ring D showed long-range connectivity with C-19, and H-18 (δ_{H} 4.51) correlated with C-16. The ³J_{CH} correlation of OCH_3 -15³ (δ_{H} 3.75) with the carboxyl group C-15² confirmed the position of the methyl-ester. The presence of the phytol-moiety attachment to the porphyrin system was indicated by the ²J_{CH} correlation of 17²-B (δ_{H} 2.22) and 17²-A (2.49) to the carboxyl function C-17³ (δ_{C} 173.30).

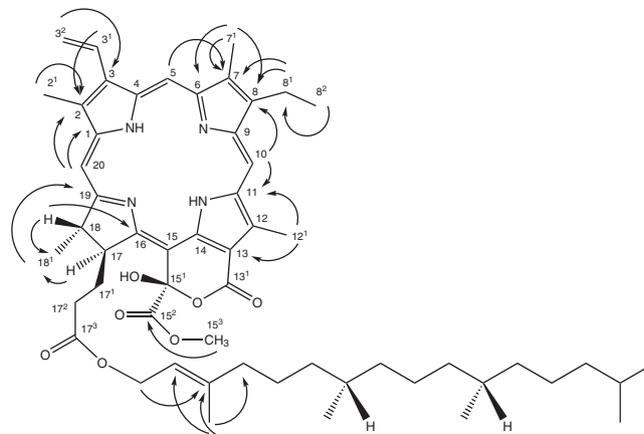


Figure 2. Structure corresponding to long-range HC-correlation signals observed in the HMBC of compound **1**

The configuration of C-15¹ was determined as *S* due up-field of H-17 (δ_{H} 4.16),³⁰ since the Nuclear Overhauser enhancements in the NOESY spectrum did not show signals referring to this chiral center. Other significant signals were observed in the porphyrin system (Figure 3). Therefore, the structure of compound **1** was identified as 15¹-hydroxy-(15¹-*S*)-porphyrinolactone (anamariaine). This is its first isolation as a natural compound.

Compound **2** displays identical signals as compared to compound **1**, except for an additional ethoxy group identified from ¹H resonances at δ_{H} 4.36 (q, $J = 7.1$ Hz, H-2-1'') and 1.50 (t, $J = 7.10$ Hz, H3-2'') and the respective carbons at δ_{C} 62.44 and 15.62. The ethoxy group was deduced as being positioned at C-15¹ on the basis of the HMBC correlation between H2-1'' and C-15¹ (δ_{C} 106.31). The spectral data of compound **2**, including the chiral center at C-15¹ (δ_{H} 4.82), agrees with those described in the literature and allows for its identification as 15¹-ethoxy-(15¹-*R*)-porphyrinolactone.³¹ This compound has been previously isolated from the green alga *Cladophora fascicularis*.³¹

The isolation of phaeophytins from *Thyracanthus ramosissimus* raises the question as to whether this material occurs naturally or is

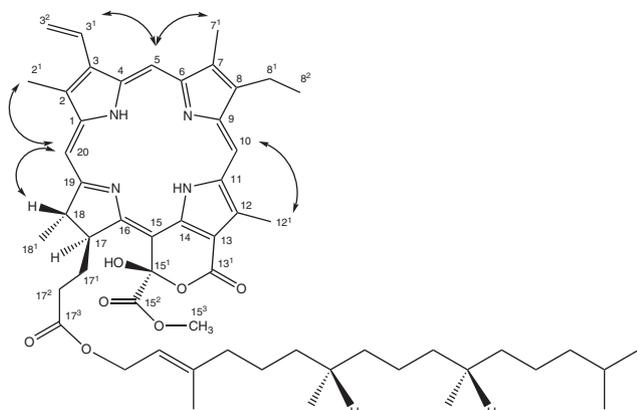


Figure 3. Structure corresponding to NOESY correlation signals observed in compound **1** showing proton-proton through-space interactions in the porphyrin-ring system

a possible artifact produced during chromatographic separation. Our experience suggests that this rare pigment is not commonly observed in plant species.^{32,33} Comparison between the structures of compounds **1** and **2** shows that an interconversion of these two pigments under the mild conditions of the isolation techniques should be highly unlikely.

The effects of the compounds on the catalytic activity of the DNA topo II- α enzyme were observed in the relaxation assays using pBR322 in the presence of ATP. Compounds **1** and **2** were evaluated at 50 μ M (lanes 4 and 5, respectively) (Figure 4). The two compounds promoted significant inhibition of the catalytic activity of topo II- α as compared to etoposide, which was used as the positive control.

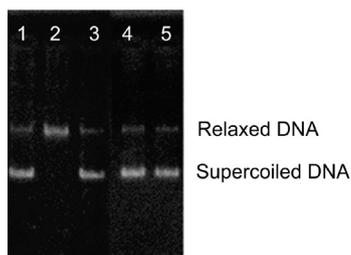


Figure 4. All lanes contain 0.152 μ g DNA (pBR322) and 1.0 unit of topo II- α , with the exception of Lane 1. Lane 1: negative control (pBR322 only). Lane 2: positive control (pBR322 and topo II- α). Lane 3: etoposide (100 mM). Lane 4: compound **1** (50 μ M). Lane 5: compound **2** (50 μ M)

SUPPLEMENTARY MATERIAL

¹H and ¹³C NMR, HMQC, COSY, HMBC, NOESY, IR, and ESI-MS spectra of compound **1** are available at <http://quimicanova.sbgq.org.br>, in PDF file, with free access.

ACKNOWLEDGEMENTS

This work was financially supported by CNPq/CAPES/FAPESQ/PRONEX. We are also extremely grateful to CENAUREM/UFC for conducting the 200 MHz spectra. The authors are also thankful to the technicians R. N. da Silva Filho (UFPB) and D. E. de A. Uchoa (UFC) for the technical support. Dr. A. Leyva helped with editing of the English language.

REFERENCES

- Barroso, G. M.; Peixoto, A. L.; Costa, C. G.; Ichaso, C. L. F.; Guimarães, E. F.; Lima, H. C.; *Sistemática das Angiospermas do Brasil*,

- Imprensa Universitária: Viçosa, 1991, vol. 3.
- Braz, D. M.; Carvalho-Okano, R. M.; Kameyama, C.; *Rev. Bras. Bot.* **2002**, *25*, 495.
- Amer, M. E.; Abou-Shoer, M. I.; Abdel-Kader, M. S.; El-Shaibany, A. M. S.; Abdel-Salama, N. A.; *J. Braz. Chem. Soc.* **2004**, *15*, 262.
- Rao, Y. K.; Vimalamma, G.; Rao, C. V.; Tzeng, Y.; *Phytochemistry* **2004**, *65*, 2317.
- Rasoamiaranjanahary, L.; Marston, A.; Guilet, D.; Schenk, K.; Randimbivololona, F.; Hostettmann, K.; *Phytochemistry* **2003**, *62*, 333.
- Leal, L. K. A. M.; Ferreira, A. A. G.; Bezerra, G. A. A.; Matos, F. J. A.; Viana, G. S. B.; *J. Ethnopharmacol.* **2000**, *70*, 151.
- Venkataraman, R.; Gopalakrishnan, S.; *Phytochemistry* **2002**, *61*, 963.
- Siripong, P.; Yahuafai, J.; Shimizu, K.; Ichikawa, K.; Yonezawa, S.; Asai, T.; Kanokmedakul, K.; Ruchirawat, S.; Oku, N.; *Biol. Pharm. Bull.* **2006**, *29*, 2279.
- Kanchanapoom, T.; Noiarsa, P.; Ruchirawat, S.; Kasai, R.; Otsuka, H.; *Phytochemistry* **2004**, *65*, 2613.
- Asano, J.; Chiba, K.; Tada, M.; Yosmil, T.; *Phytochemistry* **1996**, *42*, 713.
- Luo, Y.; Feng, C.; Tian, Y.; Zhang, G.; *Phytochemistry* **2002**, *61*, 449.
- Côrtes, A. L. A.; Borges, R. L. B.; Rapini, A.; *Taxon* **2010**, *59*, 965.
- Silva, M. N.; Ferreira, V. F.; Souza, M. C. B. V.; *Quim. Nova* **2003**, *26*, 407.
- Oliveira, M. C. C.; Carvalho, M. G.; Grynberg, N. F.; Brioso, P. S. T.; *Planta Med.* **2005**, *71*, 561.
- Cunha, A. S.; Lima, E. L. S.; Pinto, A. C.; Esteves-Souza, A.; Echevarria, A.; Câmara, C. A.; Vargas, M. D.; Torres J. C.; *J. Braz. Chem. Soc.* **2006**, *17*, 439.
- Vega, M. R. G.; Esteves-Souza, A.; Vieira, I. J. C.; Mathias, L.; Braz-Filho, R.; Echevarria, A.; *J. Braz. Chem. Soc.* **2007**, *18*, 1554.
- Esteves-Souza, A.; Figueiredo, D. V.; Esteves, A.; Câmara, C. A.; Vargas, M. D.; Pinto, A. C.; Echevarria, A.; *Braz. J. Med. Biol. Res.* **2007**, *40*, 1399.
- Branco, A.; Pinto, A. C.; Braz-Filho, R.; Silva, E. F.; Grynberg, N. F.; Echevarria, A.; *Rev. Bras. Farmacogn.* **2008**, *18*, 703.
- Cotrim, C. A.; Garrido, S. S.; Trovatti, E.; Marchetto, R.; *Quim. Nova* **2010**, *33*, 841.
- Bruxel, F.; Guterres, S. S.; Teixeira, H. F.; *Quim. Nova* **2011**, *34*, 1643.
- Corbett, A. H.; Osheroff, N.; *Chem. Res. Toxicol.* **1993**, *6*, 585.
- Nitiss, J. L.; *Biochim. Biophys. Acta* **1998**, *1400*, 63.
- Wang, J. C.; *Annu. Rev. Biochem.* **1996**, *65*, 635.
- Dias, C. S.; Moura, M. D.; Cabral, A. G. S.; Mota, S. G. R.; Cunha, E. V. L.; Silva, T. M. S.; Harley, A. M. G.; Barbosa-Filho, J. M.; *Labciencia* **2007**, *1*, 14.
- Buchanan, M. S.; Hashimoto, T.; Asakawa, Y.; *Phytochemistry* **1996**, *41*, 1373.
- Helaja, J.; Stapelbroek-Möllmann, M.; Hynninen, P. H.; *J. Org. Chem.* **2000**, *65*, 3700.
- Schwikkard, S. L.; Mulholland, D. A.; Hutchings, A.; *Phytochemistry* **1998**, *49*, 2391.
- Tomaz, A. C. A.; Nogueira, R. B. S. S.; Pinto, D. S.; Agra, M. F.; Souza, M. F. V.; Cunha, E. V. L.; *Rev. Bras. Farmacogn.* **2008**, *18*, 47.
- Jerz, G.; Arrey, T. N.; Wray, V.; Du, Q.; Winterhalter, P.; *Innov. Food Sci. Emerging Technol.* **2007**, *8*, 413.
- Nakatani, Y.; Ourisson, G.; Beck, J. P.; *Chem. Pharm. Bull.* **1981**, *29*, 2261.
- Huang, X.; Li, M.; Xu, B.; Zhu, X.; Deng, Z.; Lin, W.; *Molecules* **2007**, *12*, 582.
- Silva, T. M. S.; Câmara, C. A.; Medeiros, F. D.; Oliveira, E. J.; Agra, M. F.; Harley, R. M.; Giulietti, A. M.; *Biochem. Syst. Ecol.* **2006**, *34*, 263.
- Silva, T. M. S.; Câmara, C. A.; Barbosa-Filho, J. M.; Giulietti, A. M.; *Quim. Nova* **2010**, *33*, 571.