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Chemical Modification of Niobium Layered Oxide by Tetraalkylammonium Intercalation

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Table S1. Estimated composition of hexaniobate-tma$^+$ deposited solids

Sample	C (%)	H (%)	N (%)	Water (%)
tma(0.25)dep	2.05	1.16	0.60	4.40
$[(CH_3)_4N^+]_{0.42}H_{1.58}K_2Nb_6O_{17}\cdot 2.4H_2O$	(1.94)[a]	(1.10)	(0.48)	(4.44)
tma(0.50)dep	4.41	1.79	1.29	5.23
$[(CH_3)_4N^+]_{0.95}H_{1.05}K_2Nb_6O_{17}\cdot 3H_2O$	(4.43)	(1.80)	(1.15)	(5.38)
tma(0.75)dep	4.85	1.88	1.41	5.19
$[(CH_3)_4N^+]_{1.05}H_{0.95}K_2Nb_6O_{17}\cdot 3H_2O$	(4.83)	(1.93)	(1.24)	(5.33)
tma(1.0)dep	4.86	1.85	1.42	4.86
$[(CH_3)_4N^+]_{1.05}H_{0.95}K_2Nb_6O_{17}\cdot 2.8H_2O$	(4.90)	(1.90)	(1.16)	(4.91)

[a] experimental data.

Table S2. Estimated composition of hexaniobate-tea$^+$ deposited solids

Sample	C (%)	H (%)	N (%)	Water (%)
tea(0.25)dep	4.51	1.65	0.66	4.94
$[(CH_3CH_2)_4N^+]_{0.48}H_{1.52}K_2Nb_6O_{17}\cdot 2.8H_2O$	(4.69)[a]	(1.38)	(0.71)	(4.95)
tea(0.50)dep	5.63	2.02	0.82	6.3
$[(CH_3CH_2)_4N^+]_{0.62}H_{1.38}K_2Nb_6O_{17}\cdot 3.7H_2O$	(5.39)	(1.59)	(0.71)	(6.2)
tea(0.75)dep	5.18	1.85	0.75	5.55
$[(CH_3CH_2)_4N^+]_{0.56}H_{1.44}K_2Nb_6O_{17}\cdot 3.2H_2O$	(5.30)	(1.41)	(0.91)	(5.46)
tea(1.0)dep	5.18	1.85	0.75	5.54
$[(CH_3CH_2)_4N^+]_{0.56}H_{1.44}K_2Nb_6O_{17}\cdot 3.2H_2O$	(5.24)	(1.37)	(0.72)	(5.49)

[a] experimental data.

Table S3. Estimated composition of hexaniobate-tpa$^+$ deposited solids

Sample	C (%)	H (%)	N (%)	Water (%)
tpa(0.25)dep	1.36	0.80	0.13	3.00
$[(CH_3CH_2CH_2)_4N^+]_{0.09}H_{1.91}K_2Nb_6O_{17}\cdot 1.6H_2O$	(1.39)[a]	(0.56)	(0.03)	(3.27)
tpa(0.50)dep	3.66	1.21	0.36	2.92
$[(CH_3CH_2CH_2)_4N^+]_{0.25}H_{1.75}K_2Nb_6O_{17}\cdot 1.6H_2O$	(3.68)	(1.11)	(0.28)	(2.94)
tpa(0.75)dep	4.22	1.30	0.41	2.73
$[(CH_3CH_2CH_2)_4N^+]_{0.29}H_{1.71}K_2Nb_6O_{17}\cdot 1.5H_2O$	(4.28)	(1.16)	(0.40)	(2.78)
tpa(1.0)dep	4.48	1.36	0.44	2.89
$[(CH_3CH_2CH_2)_4N^+]_{0.31}H_{1.69}K_2Nb_6O_{17}\cdot 1.6H_2O$	(4.54)	(1.27)	(0.49)	(3.00)

[a] experimental data.

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