# *In Vitro* Screening of Marinoquinolines Compounds as Potential Antimalarial Target.

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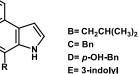
Natural isolates of Marinoquinolines, pyrroloquinoline compounds were found to have antimalarial properties. In our collaborative study, we synthesized new derivatives from MQs by *Heck-Matsuda* and *Pictet-Spengler* reaction. Further we showed that these synthesized compounds possess antimalarial properties.

## Introduction

*Plasmodium,* an etiological agent of malaria infect more than 400 million people with approximately one million global mortality rate annually. Emergence of resistant *Plasmodium* strains require urgent tackling for the development of new antimalarials.

Recently novel compounds, known as marinoquinolines A-F (MQs) were isolated from marine gliding bacteria *Rapid ithrix thailandica* and *Ohtaekwangia kribbensis*<sup>[1, 2]</sup>. It has been reported that these compounds also have antimalarial properties<sup>[3]</sup>. In this study we have synthesized 13 MQ analogues and studied their inhibitory activity against *P. falciparum* 3D7.



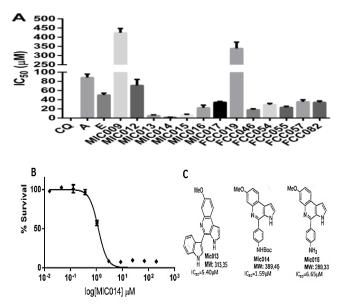


Marinoquinoline A Marinoquinoline B-F F= indole-3-carbonyl Figure 1. General structure of natural isolates of Marinoquinolines (from Schwalm C.S. *et al*).

## **Results and Discussion**

We selected different concentration range for all MQs analogues and tested their potential antimalarial properties against *P. falciparum*. Fig 2A represents the wide range of IC<sub>50</sub> values for all MQs while figure 2B represent the typical dose–response curve for one synthetic MQs. Fig 2C specifically shows the

structure of most active compounds with their respective  $\ensuremath{\mathsf{IC}_{50}}$  values



**Figure 2. (A)** *In vitro* inhibitory effect of natural and synthetic MQs. **(B)** typical dose-response curve to obtain  $IC_{50}$  for MQs on *P. falciparum*. **(C)** Structure and  $IC_{50}$  values of most effective MQs.

#### Conclusion

Our result showed that all the MQs are capable of interrupting the malaria life cycle. Among them, compounds MIC013, MIC014 and MIC015 possess lower IC<sub>50</sub> value within 1-7  $\mu$ M range.

### Acknowledgment

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<sup>3</sup>Schwalm, C.S. and C.R.D. Correia, Tetrahedron Lett., 2012. **53**(36): p.4836-4840.

<sup>&</sup>lt;sup>1</sup>Kanjana-opas, A., et al. Acta Cryst. E, 2006. **62**(7): p. o2728–o2730.

<sup>&</sup>lt;sup>2</sup>Okanya, P.W., et al. J Nat Prod, 2011. **74**(4): p. 603-8.