# Phosphates Glasses via coacervation route containing cadmium ferrite magnetic nanoparticles.

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Keywords: Coacervate, Magnetic nanoparticles, core@shell.

#### Introduction

Coacervation is a processs that makes possible to incorporate inorganic compounds, such as nanoparticles and also rare earths in the polyphosphate structure at room temperature. Thus such material can be transformed into a glass using the melting-quenching process of such precursor.<sup>1</sup> The research field using glass and glass ceramics matrices containing magnetic nanoparticles is called spin-photonics, where photons can be used to standardize magnetic medium and to study the fundamental properties of the interaction of light with magnetic materials.<sup>2,3</sup> This study aimed to prepare and characterize a phosphate based glass containing magnetic nanoparticles of cadmium ferrite via coacervation route followed by meltingquenching. The materials obtained were studied by thermal analysis, Raman Spectroscopy, UV-Vis spectrospy and transmission electron microscopy.

### **Results and Discussion**

Cadmium ferrite was prepared by the coprecipitation method.<sup>4</sup> The nanoparticles were coated with an insulating layer, forming the structure of the core@shell type. The coacervate (NaPO<sub>3</sub>-CaCl<sub>2</sub>) was obtained from the interaction between the sodium polyphosphate solution (NaPO<sub>3</sub>)<sub>n</sub> (4.0 M) with a calcium chloride solution CaCl<sub>2</sub>.2H<sub>2</sub>O (2.0 M). Three samples were prepared, one without nanoparticles, and two other containing 4% and 8 % in mass of CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub>. The glass obtained from the coacervates was produced by melting the mixture at 1000 °C and subsequent quenching at room temperature. The coacevates showed efficient dispersion of the nanoparticles. Figure 1 shows the Raman spectrum of the coacervate that presents two main vibrational modes assigned to the  $v_s(P-O-$ P) and  $\upsilon_s(P-O_t)$  in 691 and 1173 cm<sup>-1</sup>. For the coacervate glass - 4% CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub> the  $v_s$ (P-O-P) and  $\upsilon_s(P-O_t)$  appear in 694 and 1171 cm<sup>-1</sup>, respectively, while for the coacervate glass - 8% CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub> in 702 and 1164 cm<sup>-1</sup>, respectively. increasing Besides, the concentration of  $CdFe_2O_4@SiO_2$  into the coacervate leads to decreasing of the relative intensity of the two vibrational modes analyzed and a higher bandwidth, mainly for the  $u_s(P-O_t)$  mode in 1164 cm<sup>-1</sup> in the coacervate - 8% CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub> glass.



**Figure 1.** Raman Spectra for glasses whit and without magnetic nanoparticles.

Transmission electron microscopy (TEM) revealed in the Figure 2-a the presence of nanoparticles dispersed through the glass, in 2-b can observe the crystal planes of the CdFe<sub>2</sub>O<sub>4</sub> proving that there was nanoparticles in the glass. The interplanar spacing calculated from FFT (Fast Fourie trasnformer) to d<sub>311</sub> is 0,26 nm the value found in the literature is 0,259 nm.



**Figure 2.** Transmission electron microscopy of glass containing 4% of CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub>.

## Conclusions

The NaPO<sub>3</sub>-CaCl<sub>2</sub> glassy matrix obtained via coacervation proved to be an efficient host for dispersion of the magnetic nanoparticles CdFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub>, according to the Raman spectroscopy and TEM data.

#### Acknowledgments

The authors are thankful grant 2013/07793-6, São Paulo Research Foundation and CNPQ.

<sup>&</sup>lt;sup>1</sup> Dias. F, et al. Langmuir. **2005**,1776.

<sup>&</sup>lt;sup>2</sup> Sugimoto. N. J. Non-Cryst. Sol. 2008, 354.

<sup>&</sup>lt;sup>3</sup> Boeglin. C, et al. *Nature*. **2010**, 465.

<sup>&</sup>lt;sup>4</sup> Park. J.; Cheon. J. J. Am. Chem. Soc. 2016.