

Temperature influence on the synthesis of two new Hg(II) coordination polymers derived from Hg(TePy)₂

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Abstract

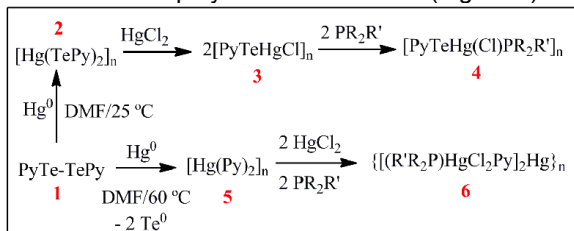
(TePy)₂ reacts with Hg⁰ and HgCl₂ at r.t. and 60 °C to form [PyTeHg(Cl)PR₂R']_n and {(R₂R'P)HgCl₂Py}₂Hg_n respectively.

Introduction

Coordination polymers are a class of compounds consisting of metal ions bonded together through multidentate organic ligands in order to form a polymeric chain. The compounds can grow from one to three dimensions. Those differences give them different properties and applications such as magnetism, photoluminescence, catalysis, molecular adsorption and sensing.^{1,2}

Results and Discussion

The oxidative addition of Hg⁰ to the bis-4-pyridyl ditelluride (**1**)³ in DMF at room temperature led to the formation of mercury (II) bis(4-pyridyltelluride) followed by the addition of HgCl₂ leading to the formation of compound **3** (Scheme 1). Then PR₂R' was added and polymer **4** was formed (Figure 1).



Scheme 1: Synthetic routes used to obtain compounds **4** and **6**. PR₂R' = P(^tBu)₂(PhNMe₂).

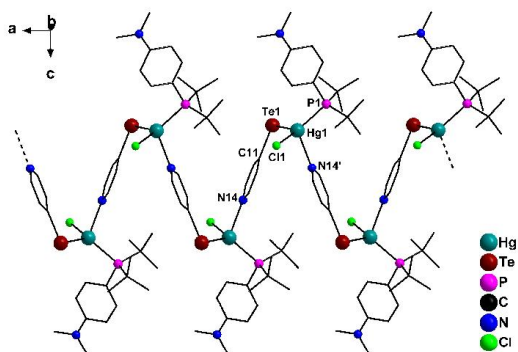


Figure 1. Polymeric structure of compound **4**.

When the bis-4-pyridyl ditelluride was carried out with Hg⁰ at 60 °C, the reductive elimination of

tellurium and oxidative addition of Hg⁰ was observed, forming the adduct [Hg(PyTe)₂]_n (**5**). With further addition of two equivalents of HgCl₂ and PR₂R' polymer **6** (Figure 2) was obtained.

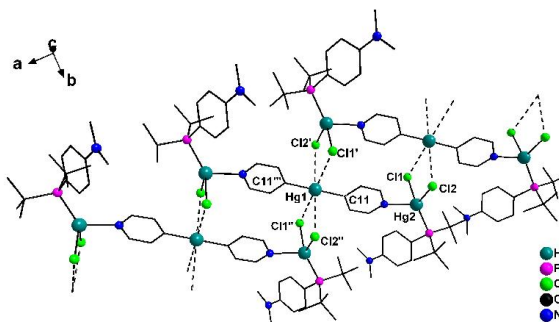


Figure 2. Polymeric structure of compound **6**.

The compounds **4** and **6** show a zigzag arrangement. Selected bond distances and angles are listed in Table 1.

Table 1. Selected bond distances (Å) and angles (°) for polymers **4** and **6**.

4	6
Te1-Hg1 = 2.691	C11-Hg1 = 2.078
Hg1-Cl1 = 2.570	C11'-Hg1 = 2.078
Hg1-P1 = 2.486	Hg1...Cl1' = 3.532
Hg1...N14' = 2.458	Hg1...Cl2' = 3.299
Te1-Hg1-P1 = 123.75	C11-Hg1-C11' = 180
Te1-Hg1-Cl1 = 113.84	Cl1'...Hg1...Cl2' = 180
Te1-Hg1-N14' = 101.06	(') = 1-x, -y, 1-z
	('') = 0.5+x, 1.5-y, 1-z

Conclusion

Two new coordination polymers based on the Hg(PyTe)₂ were synthesized and characterized using different spectroscopic techniques and X-ray crystallography. It was observed that the reaction temperature is a decisive factor for the synthesis of these compounds.

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