

# A PGSE NMR Methodology to Characterize Multi-Site Halogen-Bonded Adducts in Solution

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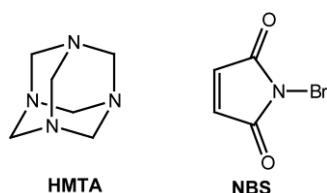
Palavras Chave: Nuclear Magnetic Resonance, Halogen Bond, Supramolecular chemistry

## Abstract

Here, a NMR method to characterize halogen-bonded adducts is presented, giving information otherwise inaccessible.

## Introdução

Halogen bonding (XB) is nowadays a powerful tool for the synthesis of advanced materials,<sup>1</sup> and, in the recent years, several applications in solution have been explored, as catalysis and anion recognition.<sup>2</sup> The characterization of XB systems in solution is often limited to titrations, through NMR (<sup>1</sup>H and <sup>19</sup>F) or UV-Vis measurements.<sup>3</sup> Recently, <sup>19</sup>F, <sup>1</sup>H HOESY NMR experiments have been used to characterize the structure of the XB adducts.<sup>4</sup> Here, we propose the diffusion NMR Pulsed field Gradient Spin Echo (PGSE)<sup>5</sup> technique as a simple and reliable tool to characterize multi-site halogen bonding systems (Scheme 1).



Scheme 1.

As known, PGSE NMR measurements give information about the hydrodynamic volume of the molecular species.<sup>5</sup> In case of an aggregation process, the measure of the hydrodynamic volume allows the quantification of the equilibrium constants involved in the process.

## Resultados e Discussão

PGSE NMR measurements on solutions containing **HMTA** (2.1 mM) and increasing concentrations of **NBS** showed that the hydrodynamic volume ( $V_H$ ) of **HMTA** increases from 217 to 438 Å<sup>3</sup>. The trend of the experimental data reveals that there is a *plateau* at high concentration of **NBS**, and  $V_H(\text{HMTA})$  tends to 450 Å<sup>3</sup>. Such a value is very similar to the

hydrodynamic volume of the 1:2 adduct. Therefore, despite **HMTA** has four different nitrogen atoms and can establish an interaction with four different molecules of **NBS**, experimental data demonstrate that the **HMTA** cannot be fully saturated and only two aggregation steps are active in solution. Fitting the experimental data considering only  $K_{a1}$  and  $K_{a2}$ , leads to  $K_{a1} = 90$  and  $K_{a2} = 60$  M<sup>-1</sup> (Figure 1).

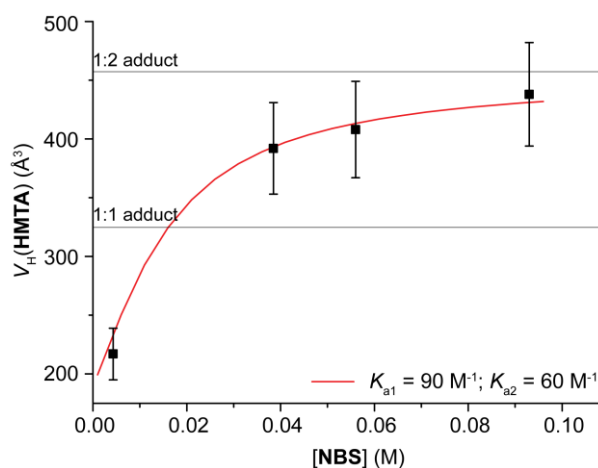


Figure 1. Experimental hydrodynamic volume of **HMTA** ( $c = 2.1$  mM) at different concentrations of **NBS**. The red line represents the best fit.

## Conclusões

In conclusion, PGSE NMR resulted to be a powerful technique to characterize multi-site XB adducts in solution, providing information about the nuclearity of the system that are very difficult to obtain with other techniques.

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