

Novel naphthoquinone-coumarin conjugates

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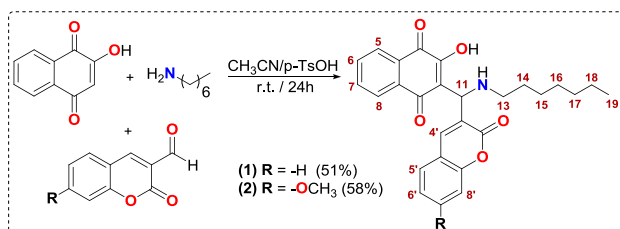
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Introduction

Mannich bases derived from lawsone are potential antineoplastic agents.¹ Besides their various biological activities,² coumarins often exhibit strong fluorescence.³ This work aims at the synthesis of novel hybrid compounds containing these two nuclei with fluorescent properties for biological activity studies. The rare examples of naphthoquinone-coumarin hybrids are natural products.⁴

Results and Discussion

Compounds **1** and **2** were obtained as orange solids from the Mannich reaction⁵ of lawsone with heptylamine and 3-formylcoumarins⁶.



Scheme 1. Synthesis of new Mannich bases.

The ¹H NMR spectra of **1** and **2** (DMSO-d₆, 500 MHz) exhibit typical signals of the naphthoquinone and coumarin aromatic hydrogens in the δ 8.2-6.9 range. The H11 signal appears around δ 5.6 and the other aliphatic protons, in the expected regions.

Table 1. UV-Vis data for **1** and **2** (MeCN, 1.64x10⁻⁵ mol/L).

	λ_1 (nm)	Log ϵ_1	λ_2 (nm)	Log ϵ_2	λ_3 (nm)	Log ϵ_3
1	271	4.43	314	3.91	438	3.27
2	270	4.38	328	4.28	439	3.33

TD-DFT calculations (B3LYP/6-31+G(d,p)/PCM) were carried out to understand the electronic structure of the conjugates. The lowest absorption band in the UV-Vis spectra, around 440 nm (see Fig. 1 and 2) is characterized by two main contributions: CT from the amino group lone electron pair (HOMO) to the naphthoquinone (LUMO), and by the transfer from the coumarin moiety (HOMO-1) to the naphthoquinone (LUMO). The band at ca. 320 nm is assigned to naphthoquinone ($\pi-\pi^*$) and the most intense band around 270 nm, to coumarin ($\pi-\pi^*$).

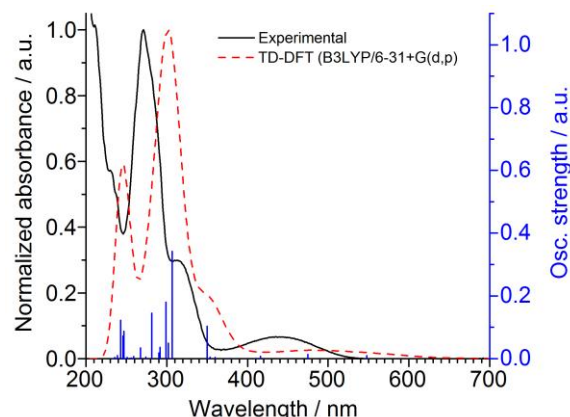


Figure 1. Experimental and theoretical UV-Vis absorption spectra for **1** in MeCN.

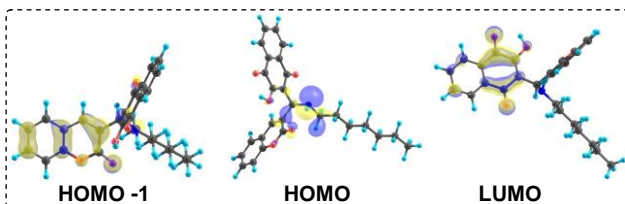


Figure 2. Molecular orbitals involved in the electronic charge transfer.

Preliminary fluorescence data were obtained (MeCN) by excitation of the three absorption bands (λ_{max} , Table 1) of all compounds. Compounds **1** and **2**, when excited at λ_1 , have emission maxima around 310 nm. Excitation of λ_2 resulted in emissions at 348 (**1**) and 430 nm (**2**), while excitation of the lowest energy band λ_3 resulted in emissions at 582 (**1**) and 573 nm (**2**).

Conclusions

Two novel naphthoquinone-coumarin conjugates were synthesized and characterized by ¹H NMR and UV-visible spectroscopy. Their electronic structures were characterized as a donor-acceptor system. Additional fluorescence studies are in progress.

Acknowledgements

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¹ Neves, A. P. et al. *Dalton Trans.* **2010**, 39, 10203.

² Serra, S. et al. *Bioorg. Med. Chem. Lett.* **2012**, 22, 258.

³ Zhou, Z.; Li, N.; Tong, A. *Anal. Chim. Acta.* **2011**, 702, 81.

⁴ Miranda, F.S. et al. *J. Braz. Chem. Soc.* **2014**, 25, 133.

⁵ Fiorot, R. G. et al. *Tetrahedron Letters* **2014**, 55, 4373-4377.

⁶ René, L.; Lefebvre, A.; Auzou, G. *Synthesis* **1986**, 7, 567.